organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

G. Y. S. K. Swamy,^a* K. Ravikumar,^a P. Narender^b and V. Jayathirtha Rao^b

^aLaboratory of X-ray Crystallography, Indian Institute of Chemical Technology, Hyderabad 500 007, India, and ^bOrganic Chemistry Division II, Indian Institute of Chemical Technology, Hyderabad 500 007, India

Correspondence e-mail: swamy@ins.iictnet.com

Key indicators

Single-crystal X-ray study T = 273 K Mean σ (C–C) = 0.002 Å R factor = 0.041 wR factor = 0.127 Data-to-parameter ratio = 19.6

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

2-[(2-Chloro-5-ethyl-3-pyridinyl)(hydroxy)methyl]acrylonitrile

In the title compound, $C_{11}H_{11}ClN_2O$, molecules are linked *via* $O-H\cdots N$ hydrogen bonds, forming infinite chains running along the *c* axis with a graph-set motif of *C*(6). The structure is further stabilized by weak $\pi-\pi$ and $C-H\cdots\pi$ interactions.

Received 31 January 2005 Accepted 9 February 2005 Online 19 February 2005

Comment

Baylis–Hillman adducts are well known in organic synthesis because of their biological relevance (Kabat *et al.*, 1996; Kim *et al.*, 2002; Shi *et al.*, 2002). We report here the crystal structure of the title compound, (I), which is a Baylis–Hillman product.



The molecular configuration of (I) is shown in Fig. 1. The molecule contains a chiral atom (C9), although it belongs to a centrosymmetric space group and the compound is thus a racemic mixture. The least-squares plane containing the ethyl group is nearly perpendicular [84.6 (2)°] to the least-squares plane of the pyridine ring. The hydroxy O atom is tilted away from the acrylonitrile group as the C10–C9–O–1 bond angle is distinctly larger than that of C1–C9–O1 (Table 1).



© 2005 International Union of Crystallography Printed in Great Britain – all rights reserved A view of (I), with the atom-numbering scheme. Displacement ellipsoids are drawn at the 30% probability level.



Figure 2

A partial packing diagram of (I), showing the $O-H \cdots N$ hydrogenbonded (dashed lines) molecules forming chains along the *c* axis.

The crystal packing of (I) is dictated by intermolecular O– H···N hydrogen bonds (Table 2), forming chains [C(6) type; Bernstein *et al.*, 1995] along the *c* axis. An intramolecular C– H···O interaction is also observed. This interaction closes the five-membered pseudo-ring O1–C9–C1–C6–H6 according to an S(5) pattern. Similar interactions have been reported in the literature (Pálinkó, 1999). Additionally, a weak C–H··· π contact exists, involving the π -system of the pyridyl ring (Spek, 2003). The crystal structure is further stabilized by π – π interactions (Steed & Atwood, 2000), with a centre-to-centre distance of 3.654 Å.

Experimental

Compound (I) was prepared by the coupling of 2-chloro-5-ethylpyridine-3-carbaldehyde (5 mmol) and acrylonitrile (5 mmol) in methanol, the reaction mixture being stirred at room temperature in the presence of 1,4-diazabicyclo[2.2.2]octane (5 mmol) for 15 min. This mixture was washed with water, extracted with chloroform and recrystallized from acetonitrile (yield 96%).

Crystal data

Mo K α radiation Cell parameters from 9355 reflections $\theta = 4.6-55.8^{\circ}$ $\mu = 0.31 \text{ mm}^{-1}$ T = 273 (2) K Block, colourless $0.22 \times 0.18 \times 0.16 \text{ mm}$

Data collection

Bruker SMART APEX CCD area- detector diffractometer ω scans Absorption correction: none 18128 measured reflections 2699 independent reflections	2305 reflections with $I > 2\sigma(I)$ $R_{int} = 0.020$ $\theta_{max} = 28.0^{\circ}$ $h = -16 \rightarrow 16$ $k = -16 \rightarrow 16$ $l = -18 \rightarrow 18$
Refinement	
Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.041$ $wR(F^2) = 0.127$ S = 1.05 2699 reflections	$w = 1/[\sigma^{2}(F_{o}^{2}) + (0.072P)^{2} + 0.603P]$ where $P = (F_{o}^{2} + 2F_{c}^{2})/3$ $(\Delta/\sigma)_{\text{max}} < 0.001$ $\Delta\rho_{\text{max}} = 0.32 \text{ e} \text{ Å}^{-3}$

Table 1

138 parameters

Selected geometric parameters (Å, °).

H-atom parameters constrained

$\overline{C^2 - N^1}$	1 3203 (17)	C9-01	1 4057 (17)
C2-Cl C4-N1	1.7349(14) 1.337(2)	C12-N2	1.142 (3)
O1-C9-C1	107.51 (11)	O1-C9-C10	109.90 (12)

 $\Delta \rho_{\rm min} = -0.27 \ {\rm e} \ {\rm \AA}^{-3}$

Table 2Hydrogen-bonding geometry (Å, $^{\circ}$).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$O1-H1\cdots N1^{i}$	0.82	2.00	2.8123 (15)	174
C6−H6···O1	0.93	2.31	2.6646 (16)	102
$C8-H8C\cdots Cg^{ii}$	0.96	2.95	3.581 (4)	124

Symmetry codes: (i) $\frac{1}{2} - x$, y, $z - \frac{1}{2}$; (ii) -x, -y, -z.

Note: Cg is the centroid of the pyridyl ring.

All H atoms were placed in geometrically idealized positions and allowed to ride on their parent atoms [C-H = 0.93-0.98 Å and O-H = 0.82 Å, and $U_{iso}(H) = 1.5U_{eq}(C,O)$ for methyl and hydroxy H atoms and $1.2U_{eq}(C)$ for all other H atoms].

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINT* (Bruker, 2001); data reduction: *SAINT*; structure solution: *SHELXS97* (Sheldrick, 1997); structure refinement: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3* (Farrugia, 1997) and *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97* and *PARST* (Nardelli, 1995).

The authors thank Dr J. S. Yadav, Director, IICT, Hyderabad, for his kind encouragement.

References

Bernstein, J., Davis, R. E., Shimoni, L. & Chang, N.-L. (1995). Angew. Chem. Int. Ed. Engl. 34, 1555–1573.

Bruker (2001). SAINT (Version 6.28a) and SMART (Version 5.625). Bruker AXS Inc., Madison, Wisconsin, USA.

Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.

Kabat, M. M., Kiegel, J., Cohen, N., Toth, K., Wovkulich, P. M. & Uskokovic, M. R. (1996). J. Org. Chem. 61, 118–124.

Kim, J. N., Lee, H. J. & Gong, J. H. (2002). *Tetrahedron Lett.* 43, 9141–9146. Nardelli, M. (1995). *J. Appl. Cryst.* 28, 659.

- Pálinkó, I. (1999). Acta Cryst. B55, 216-220.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.

Shi, M., Zhao, G. L. & Wu, X. F. (2002). Eur. J. Org. Chem. pp. 3666-3679.

Spek, A. L. (2003). J. Appl. Cryst. 36, 7-13.

Steed, J. W. & Atwood, J. L. (2000). Supramolecular Chemistry. Chichester: Wiley.